

**Paper Reference(s) 1CH0/2F**  
**Pearson Edexcel Level 1/Level 2 GCSE**  
**(9–1)**

**Chemistry**  
**PAPER 2:**  
**Foundation Tier**

**Diagram Booklet**

**In the boxes below, write your name, centre number and candidate number.**

<b>Surname</b>					
<b>Other names</b>					
<b>Centre Number</b>					
<b>Candidate Number</b>					

## **INSTRUCTIONS**

**There may be spare copies of some diagrams in case you need them.**

**THIS DIAGRAM BOOKLET MUST BE  
RETURNED WITH THE QUESTION PAPER  
AT THE END OF THE EXAMINATION.**

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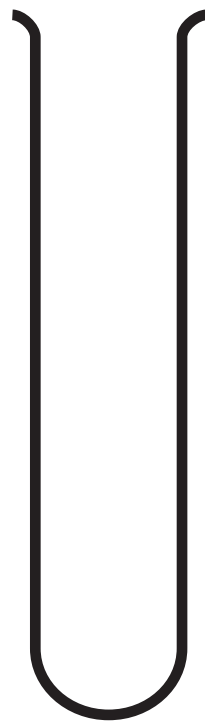
**4**  
**Question 1(d)**

**FIGURE 1**

<b>substance</b>	<b>percentage composition</b>
<b>abrasives</b>	<b>35%</b>
<b>water</b>	
<b>other substances</b>	<b>25%</b>

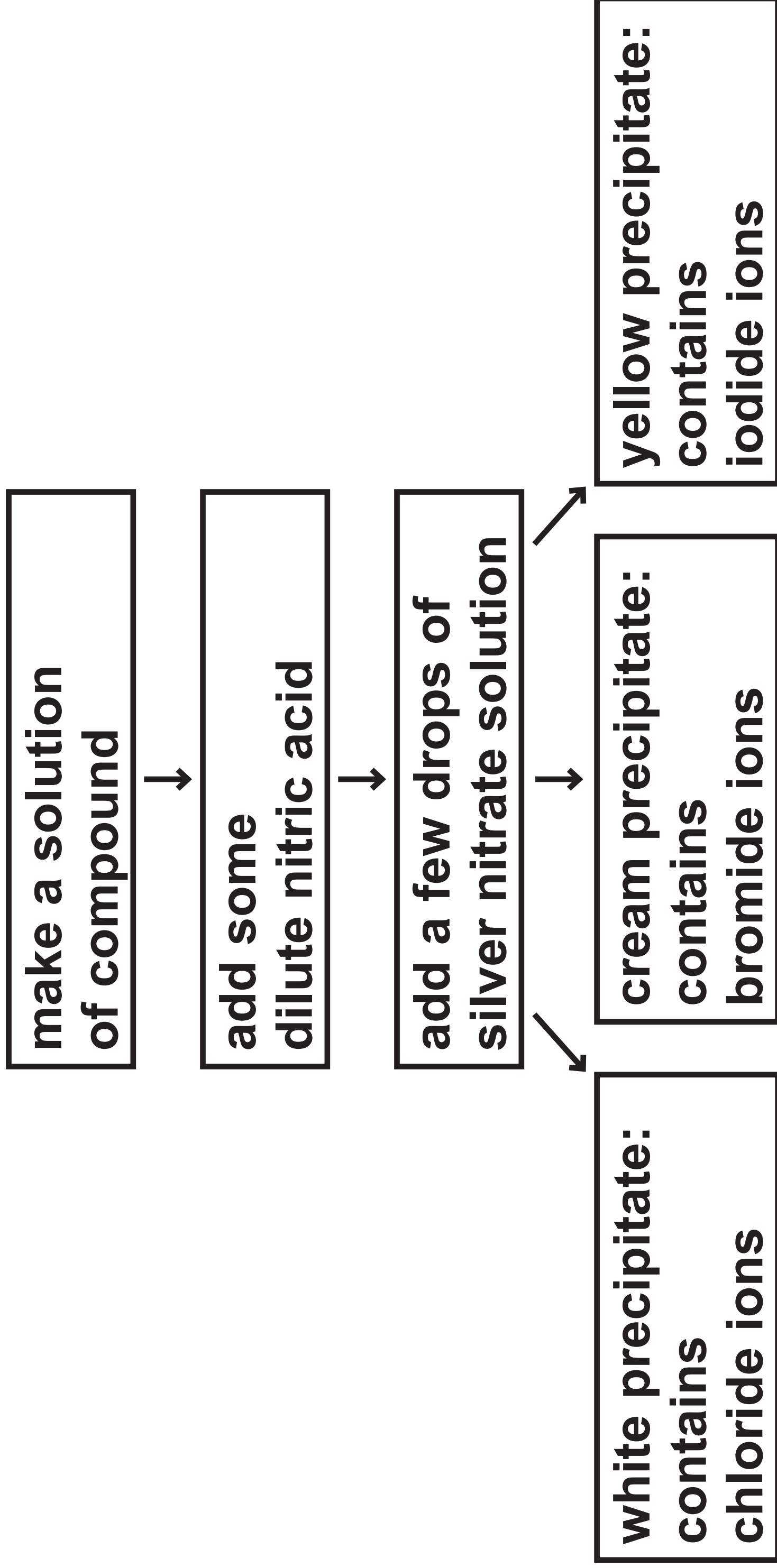
**Question 2(a)****FIGURE 2**

<b>symbol</b>	<b>melting point in °C</b>
<b>Li</b>	<b>181</b>
<b>Na</b>	<b>98</b>
<b>K</b>	<b>64</b>

**Question 2(b)****FIGURE 3**

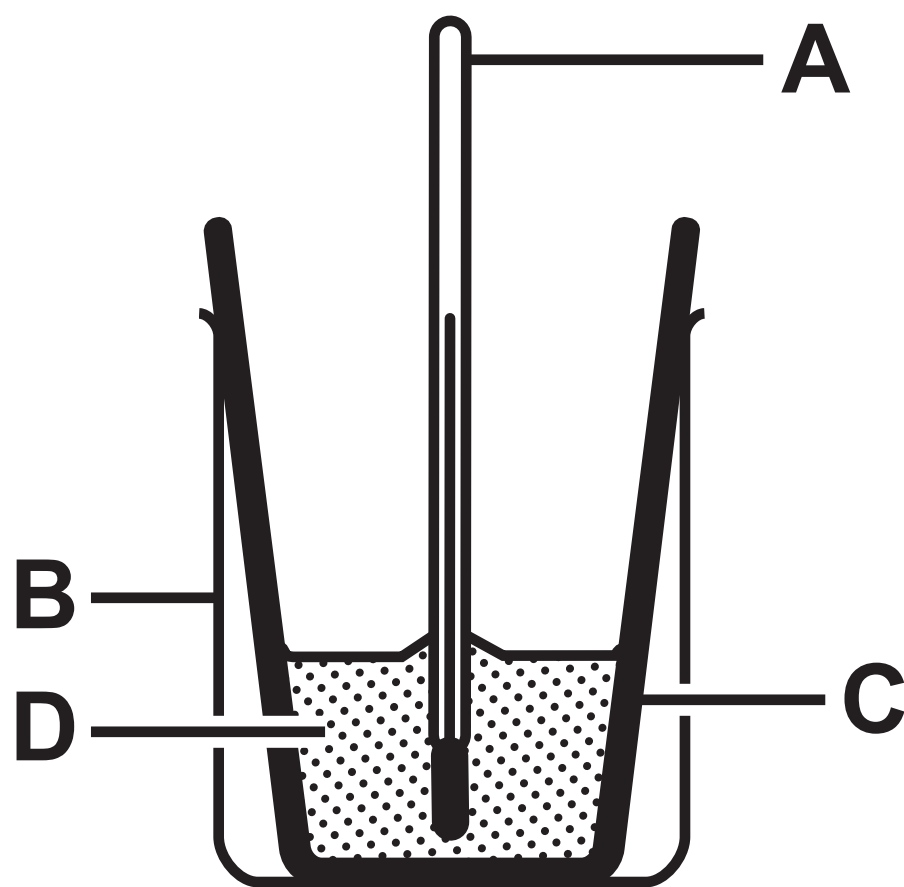
**7**  
**Question 3**

**FIGURE 4**



## Question 4(b)

FIGURE 5





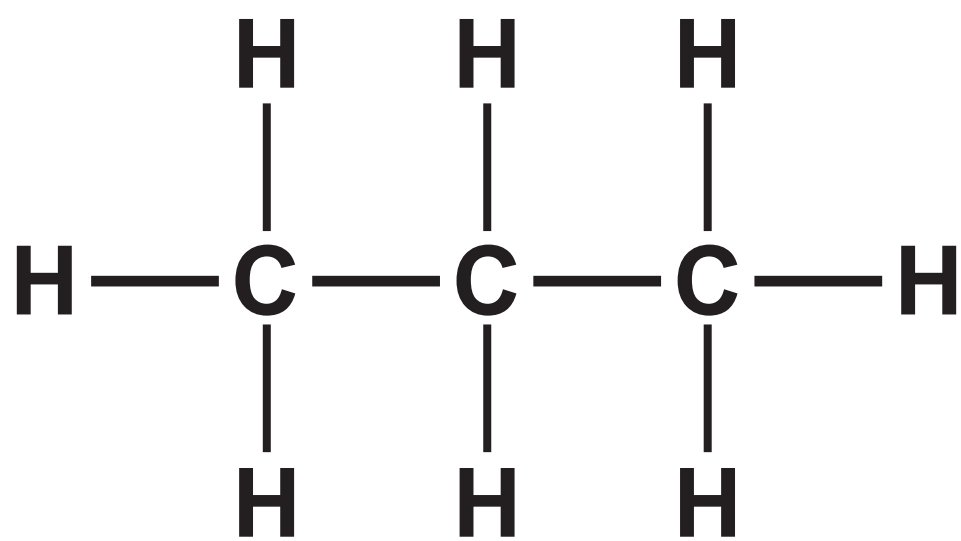
Question 4(b)(iv)

FIGURE 6

temperature of liquid at start in °C	18.6
temperature of products at end in °C	16.1

## Question 5(a)

FIGURE 7



## Question 5(b)

**fraction****use****petrol**

- **fuel for aircraft**

- **fuel for ships**

**kerosene**

- **fuel for cars**

- **making plastic**

**bitumen**

- **extracting iron**

- **making road surfaces**

## Question 5(b)

**fraction****use****petrol**

- **fuel for aircraft**

- **fuel for ships**

**kerosene**

- **fuel for cars**

- **making plastic**

**bitumen**

- **extracting iron**

- **making road surfaces**

## Question 6(b)

FIGURE 8

<b>halogen</b>	<b>description of reaction with heated iron wool</b>
<b>bromine</b>	<b>reacts quickly</b>
<b>chlorine</b>	<b>reacts very quickly</b>
<b>iodine</b>	<b>reacts slowly</b>

**Question 6(c)**

**an acid**

**a catalyst**

**higher**

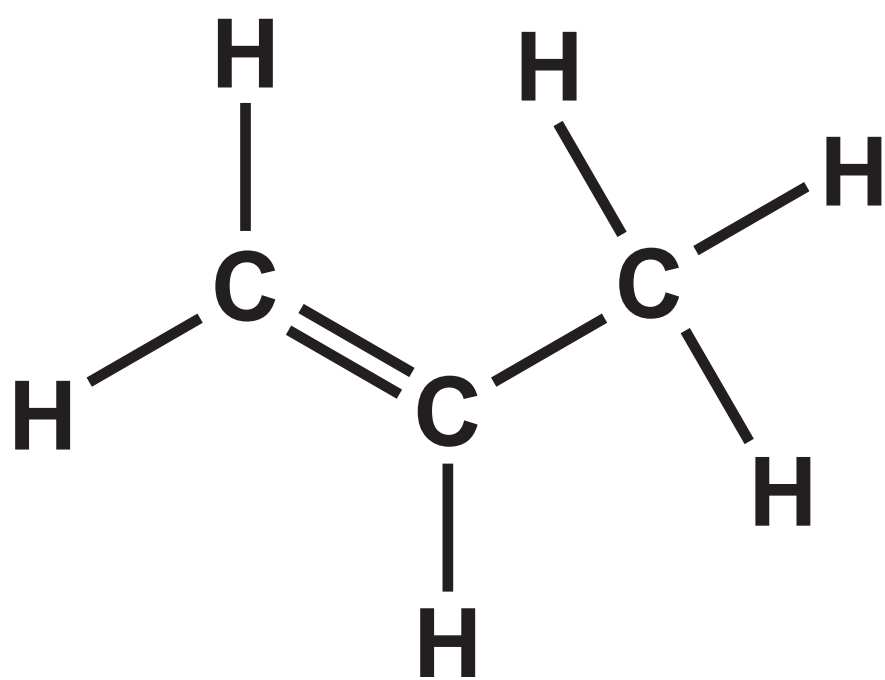
**lower**

**a reactant**

**unchanged**

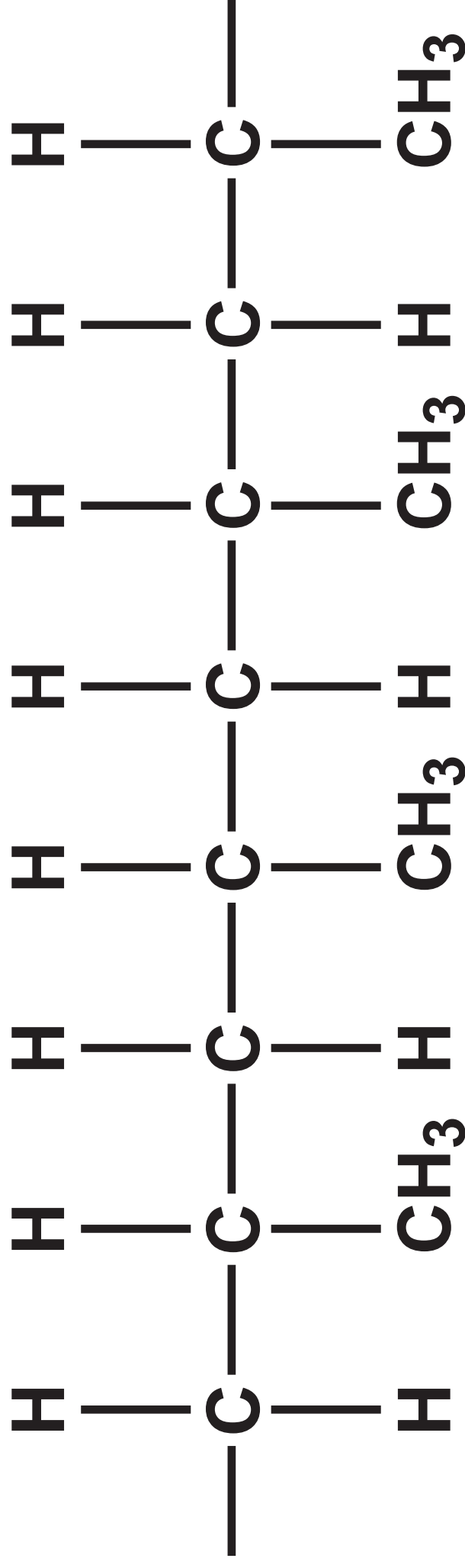
## Question 7

FIGURE 9



## Question 7(c)

FIGURE 10





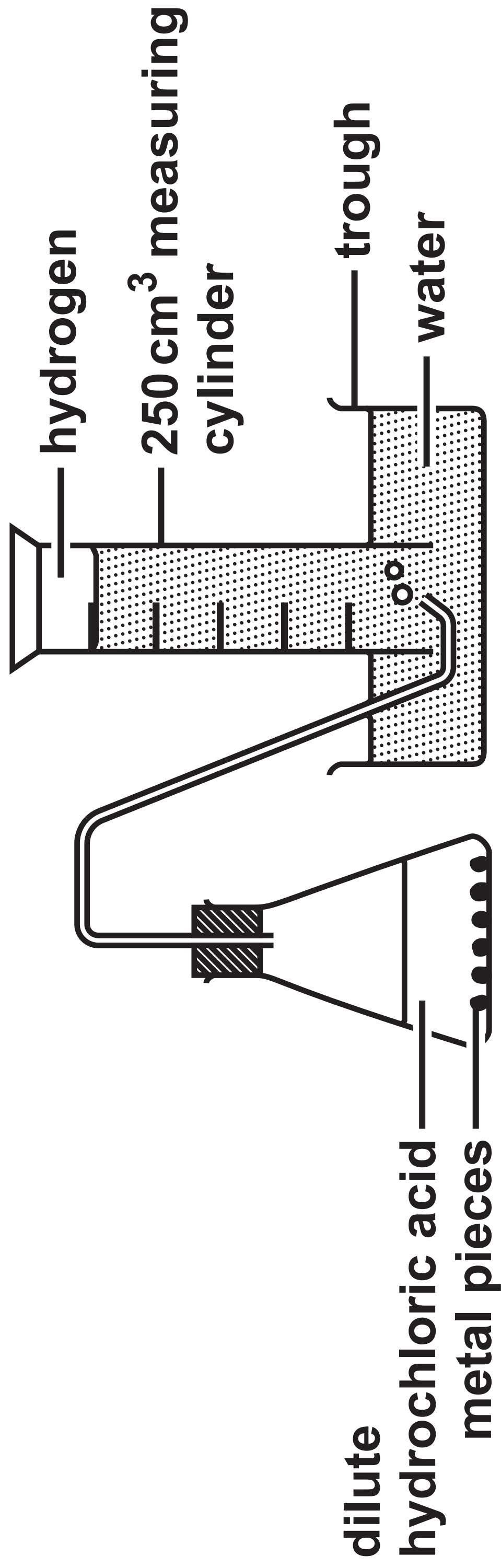
## Question 7(d)

FIGURE 11

alkane	temperature change in °C
methane	9
ethane	16
propane	22
butane	29

18  
Question 8

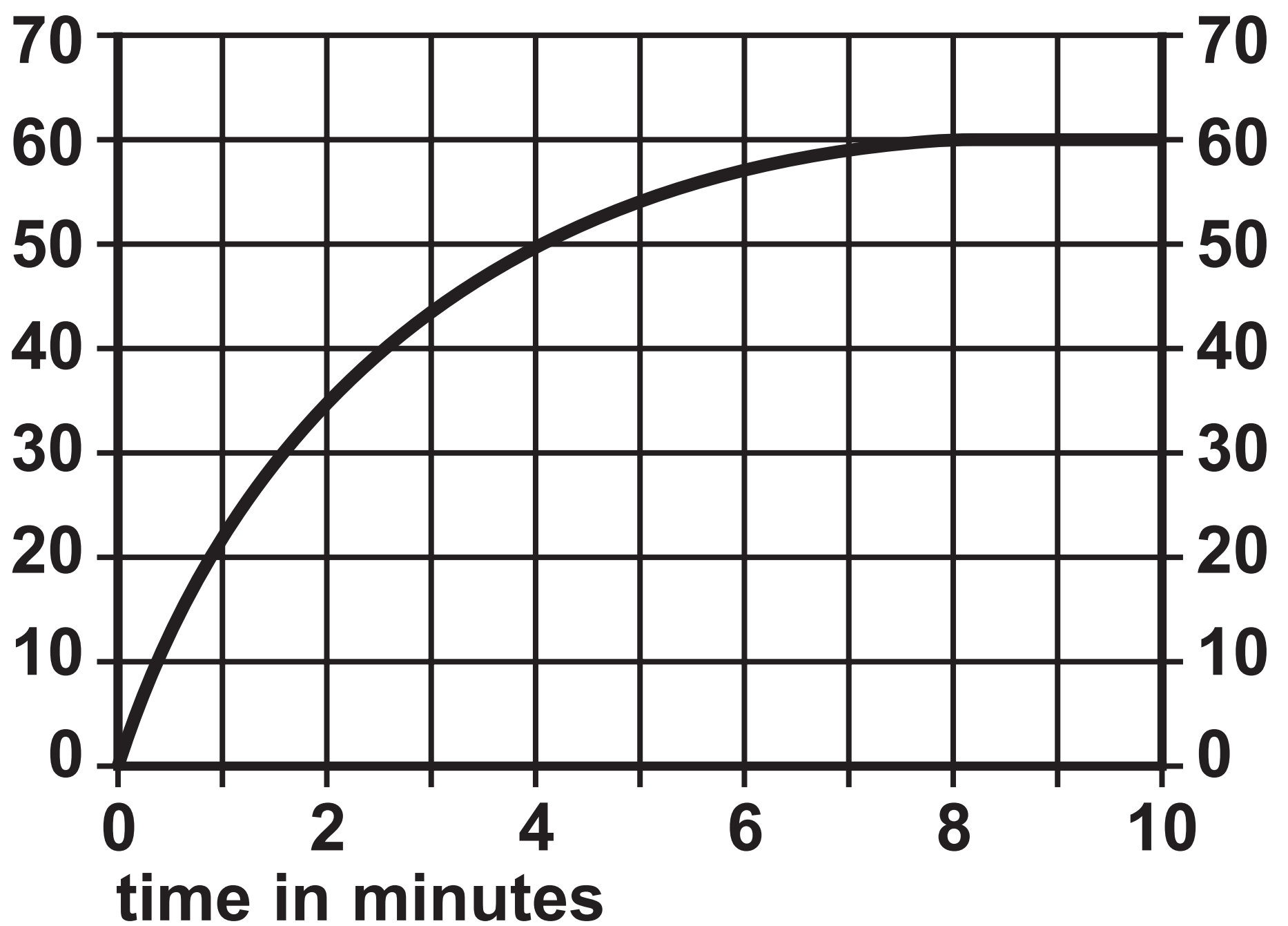
FIGURE 12



## Question 8(a)

FIGURE 13

volume of  
hydrogen in  $\text{cm}^3$



## Question 9(d)

FIGURE 14

## Key

⊖ = electron

● = neutron

⊕ = proton

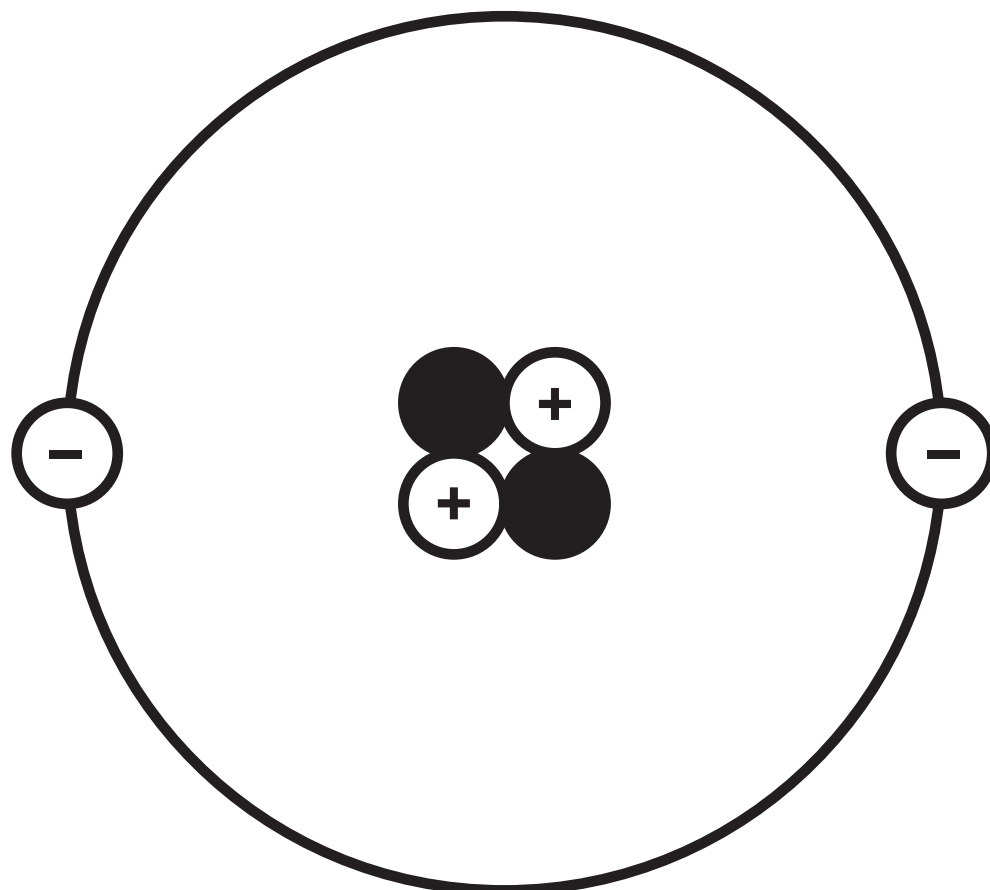


FIGURE 15

gas	relative amount in early atmosphere	composition of today's atmosphere
water vapour	large amount	0 % to 4 %
carbon dioxide	large amount	less than 0-5 %
oxygen	little or none	21 %

## Question 10(a)

FIGURE 16



**Question 10(b)(ii)****FIGURE 17**

<b>compound</b>	<b>flame colour</b>
<b>P</b>	<b>red</b>
<b>Q</b>	<b>lilac</b>
<b>R</b>	<b>blue-green</b>